The following section provides a very brief introduction to this topic, which is frequently encountered as part of addressing specific impacts within estuaries. The section is included so that these aspects are not forgotten but the reader is referred to the literature cited for more extensive coverage.

SEDIMENT QUALITY

Naturally occurring and synthetic contaminants such as heavy metals, hydrocarbons, organotins (TBT), pesticides, organochlorides, organophosphates, solvents and polychlorinated biphenyls (PCBs) tend to be associated with fine sediments (silts and clay) due in part to their greater surface area for geochemical binding.

The re-suspension and bioavailability of sediment bound contaminants in the water column is often an important consideration by regulators and statutory agencies when determining the acceptability of capital and maintenance dredging operations. The risks associated with such operations may be a determining factor when stipulating developmental controls such as seasonal restrictions and monitoring thresholds.

The potential risks for marine environmental receptors (biology and water quality) can be inferred initially through comparison with sediment quality standards. In the absence of statutory sediment quality values that define thresholds for acceptability (for licensed disposal for instance) the quality of material for dredge and disposal operations is typically assessed through comparison with Dutch (IADC/CEDA, 1997) and Canadian Standards (CCME, 1999).

Whilst such standards are based on biological effects for a range of organisms, the testing conditions used to derive the standards are unlikely to be representative of different dredging and construction techniques in the field. In particular, the degree of re-suspension of host suspended sediments is likely to vary.

Understanding particle-water interactions provides a means of estimating the degree of contamination of the water column (near and far-field effects) through the process of geochemical modelling.

Particle-Water Interactions

Interactions of chemical constituents between suspended particles and water result from a variety of physical, chemical and biological processes, including colloid aggregation, ion exchange, adsorption-desorption, absorption, precipitation-dissolution, hydrophobic bonding, sating out, microbiological activity and degradation of particulate organic matter.

When a contaminant is associated with the solid phase, it is not known which of these processes dominates and the transfer between solid and aqueous phase effected by chemical processes is defined by the general term sorption (Turner & Millward, 2002).

Sorption is frequently quantified by the empirically derived partition or distribution coefficient, ‘$K_d$’, which defines the concentration ratio of chemical sorbed to the different components of suspended particles (solid phase), $C_s$, to chemical dissolved in water (aqueous phase), $C_w$:

$$K_d = \frac{C_s}{C_w}$$

Values for $K_d$ vary greatly between contaminants and also as a function of aqueous and solid phase chemistry.
Sorption Models
Information on sorption models is provided in detail in (USEPA, 1999a; USEPA, 1999b) and is summarised here.

**Constant Partition Coefficient (Kd) Model**
The constant partition coefficient model, $K_d$, attempts to account for various chemical and physical mechanisms that are influenced by a range of variables, including, sediment composition and concentration, contaminant concentration, salinity and pH together with other complex chemical/physical interactions.

Given the range of influencing factors on partitioning, site-specific Kd values would ideally be needed for the system to be modeled.

**Parametric K\(_d\) Model**
A limitation of using $K_d$ values is that they are representative for only one set of environmental conditions.

In the Parametric $K_d$ model, the partition coefficient varies as a function of empirically derived relationships with aqueous and solid phase independent parameters. Thus, it has the distinct advantage of being more robust and removes the burden of determining new $K_d$ values for each environmental condition.

**Isotherm Adsorption Models**
The Isotherm model can be applied where sorption relationships deviate from linearity. Since $K_d$ is a function of contaminant concentration, the effect of contaminant concentration on adsorption can be tested, while other parameters are held constant. This results in a series of adsorption isotherms.

**Applications**
With knowledge of the partition coefficient (including uncertainty) for a particular contaminant, it is possible to apply project specific information on suspended sediment concentrations from for example, dredging activities, to estimate the concentration of contaminant released into the water column.

This approach can be applied to examine the effects on water quality in the near field (close to the source), where deterioration in water quality will primarily be a function of contaminants associated with the solid phase, and far field, where there is dispersion of aqueous phase contaminants and the risk of exceeding the relevant marine Environmental Quality Standards (EQS).

Numerical modelling can support these assessments by linking both local hydrodynamics and sediment dispersion with water quality parameters in a computational domain (see also water quality).
Analysis and Modelling Guide

Relationship Between Sediment Transport and Contaminant Transport
Given the affinity between fine sediments and contaminants, knowledge of sediment transport pathways may allow identification of sinks for contaminated sediments. Likewise, mapping of contamination in an estuary may assist the identification of sediment transport pathways.

The method of Sediment Trends Analysis (McLaren & Bowles, 1985) provides one method for testing this relationship. Sediment Trend Analysis attempts to determine the patterns of net sediment transport over an area through the grain-size distributions of the sediments (McLaren & Bowles, 1985). The sampled sediments are described in statistical terms (by the moment measures of mean, sorting and skewness) and the basic underlying assumption is that processes causing sediment transport will affect the statistics of the sediments in a predictable way. Following on from this assumption, the size frequency distributions of the sediments provide the data with which to search for patterns of net sediment transport.

When used in conjunction with sampling of the sediments for contamination, these methods have been applied to examine potential sites for dredge material disposal and environmental monitoring (McLaren, 1987; McLaren et al., 1993; McLaren, 1993).

References


McLaren P, 1993, Sediment transport in vancouver harbour: implications to the fate of contaminated sediments and/or dredged material disposal, GeoSea Consulting (Canada) Ltd, Vancouver Island, 1-25+figs.


USEPA, 1999a, Partition coefficients for metals in soils, water and waste (Draft), United States Environmental Protection Agency, Office of Solid Waste.

USEPA, 1999b, Understanding Variation in Partition Coefficient (Kd) Values, United States Environmental Protection Agency, Office of Air and Radiation, Report No: EPA 402-R-99-004A.